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### Synthesis and Magnetism Of New Bimetallic Assemblies, [Ni(diamine)<sub>2</sub>]<sub>2</sub>[Fe(CN)<sub>6</sub>]X, Extended by Fe<sup>III</sup>-CN-Ni<sup>II</sup> Linkages

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## SYNTHESIS AND MAGNETISM OF NEW BIMETALLIC ASSEMBLIES, $[\text{Ni}(\text{diamine})_2]_2[\text{Fe}(\text{CN})_6]\text{X}$ , EXTENDED BY $\text{Fe}^{\text{III}}\text{-CN-Ni}^{\text{II}}$ LINKAGES

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**Abstract** Bimetallic assemblies of the general formula,  $[\text{Ni}(\text{diamine})_2]_2\text{-}[\text{Fe}(\text{CN})_6]\text{X}$  (diamine = en, pn, 1,1-dmen (1,1-dimethylethylenediamine);  $\text{X}^- = \text{ClO}_4^-, \text{BF}_4^-$  or  $\text{PF}_6^-$ ) have been prepared as anhydrides for the en compounds and di- or trihydrates for the pn and 1,1-dmen compounds. The X-ray crystallography for  $[\text{Ni}(\text{pn})_2]_2[\text{Fe}(\text{CN})_6]\text{ClO}_4 \cdot 2\text{H}_2\text{O}$  has proved a 2-D network of a square structure extended by Fe-CN-Ni linkages. All the compounds show a ferromagnetic spin-exchange interaction between the adjacent  $\text{Fe}(\text{III})$  and  $\text{Ni}(\text{II})$  ions through cyanide bridge. The pn and 1,1-dmen compounds show a long-range magnetic ordering over the lattice (metamagnet). The en compounds probably have a network differing from the 2-D network of the pn and 1,1-dmen compounds and show no magnetic ordering over the lattice.

## INTRODUCTION

Recently there has been an increasing interest in the design of molecular-based magnetic material using paramagnetic complexes as constituents.<sup>1-3</sup> Bimetallic assemblies derived from hexacyanometallate ions  $[\text{M}(\text{CN})_6]^{n-}$  and simple metal ions form a family of magnetic materials exhibiting spontaneous magnetization and high  $T_C$  or  $T_N$  is reported for some of the assemblies such as  $\text{CsMnCr}(\text{CN})_6$  ( $T_C = 90 \text{ K}$ )<sup>4</sup>,  $\text{CsNiCr}(\text{CN})_6$  ( $T_N = 90 \text{ K}$ )<sup>5</sup>,  $\text{Cs}_2\text{MnV}(\text{CN})_6$  ( $T_N = 125 \text{ K}$ )<sup>6</sup>,  $\text{Cs}_{0.75}[\text{Cr}^{\text{II}}_{1.125}\text{Cr}^{\text{III}}(\text{CN})_6]$  ( $T_C = 190 \text{ K}$ )<sup>7</sup>,  $\text{Cr}^{\text{II}}_3[\text{Cr}^{\text{III}}(\text{CN})_6]_2$  ( $T_C = 240 \text{ K}$ ).<sup>7</sup> However, magnetostructural correlation for those high  $T_C$  or  $T_N$  compounds remains to be studied. With the aim to gain an insight into the origin for high  $T_C$  or  $T_N$  of those bimetallic assemblies, we have initiated the study on bimetallic assemblies derived from  $[\text{M}(\text{CN})_6]^{n-}$  and bis(diamine)nickel(II)  $[\text{Ni}(\text{diamine})_2]^{2+}$  complexes. In the previous communications we have reported a 1-D chain compound of a rope-ladder structure  $[\text{Ni}(\text{en})_2]_3[\text{Fe}(\text{CN})_6]_2 \cdot 2\text{H}_2\text{O}$ <sup>8</sup> and a 2-D compound of a square structure  $[\text{Ni}(\text{pn})_2]_2\text{-}[\text{Fe}(\text{CN})_6]\text{ClO}_4 \cdot 2\text{H}_2\text{O}$ .<sup>9</sup> Here we report the synthesis and magnetism of the bimetallic assemblies of the general formula  $[\text{Ni}(\text{diamine})_2]_2[\text{Fe}(\text{CN})_6]\text{X}$  (diamine = en, pn or 1,1-dmen (1,1-dimethylethylenediamine);  $\text{X}^- = \text{ClO}_4^-, \text{BF}_4^-$  or  $\text{PF}_6^-$ ).

## PREPARATIONS AND GENERAL CHARACTERIZATION

In our preliminary study the bimetallic assemblies were obtained as microcrystals by the 1 : 2 reaction of  $[\text{Ni}(\text{diamine})_2]\text{X}_2$  and  $\text{K}_3[\text{Fe}(\text{CN})_6]$  in an aqueous solution, but eventually it was found that the use of  $[\text{Ni}(\text{diamine})_3]\text{X}_2$  gave the same compounds as large crystals. It is likely that the slow dissociation of  $[\text{Ni}(\text{diamine})_3]^{2+}$  into  $[\text{Ni}(\text{diamine})_2]^{2+}$  in aqueous solution leads to growing large crystals of ordered network structure. By this method the following bimetallic assemblies have been obtained:  $[\text{Ni}(\text{en})_2]_2[\text{Fe}(\text{CN})_6]\text{X}$  ( $\text{X}^- = \text{ClO}_4$  (1),  $\text{BF}_4$  (2),  $\text{PF}_6$  (3)),  $[\text{Ni}(\text{pn})_2]_2[\text{Fe}(\text{CN})_6]\text{X} \cdot 2\text{H}_2\text{O}$  ( $\text{X}^- = \text{ClO}_4$  (4),  $\text{BF}_4$  (5),  $\text{PF}_6$  (6)), and  $[\text{Ni}(1,1\text{-dmen})_2]_2[\text{Fe}(\text{CN})_6]\text{X} \cdot n\text{H}_2\text{O}$  ( $\text{X}^- = \text{ClO}_4$  and  $n = 2$  (7),  $\text{X}^- = \text{BF}_4$  and  $n = 3$  (8),  $\text{X}^- = \text{PF}_6$  and  $n = 2$  (9)).

It must be emphasized that the en compounds (1-3) were obtained as anhydrides whereas the pn (4-6) and 1,1-dmen (7-9) compounds as hydrates ( $n = 2 - 3$ ). This fact suggests that the network structure of the en compounds differs from that of the pn and 1,1-dmen complexes. The en compounds show three  $\nu(\text{CN})$  modes of cyanide group around 2140, 2130 and 2110  $\text{cm}^{-1}$ . The vibration at 2110  $\text{cm}^{-1}$ , well corresponding to the  $\nu(\text{CN})$  band of  $\text{K}_3[\text{Fe}(\text{CN})_6]$  (2110  $\text{cm}^{-1}$ ), may be assigned to the cyanide group free from bridging to  $[\text{Ni}(\text{en})_2]^{2+}$ . It is noticed that in  $\nu(\text{CN})$  modes the en compounds resemble the  $\text{Ni}_3\text{Fe}_2$ -type bimetallic assembly having a 1-D network of a rope-ladder structure.<sup>8</sup> Such IR spectral feature of the en complexes suggests two types of cyanide bridge existing in their network. The pn and 1,1-dmen complexes show two  $\nu(\text{CN})$  modes near 2140 and 2110  $\text{cm}^{-1}$ , which are assigned to the bridging and non-bridging cyanide groups, respectively. The 2140  $\text{cm}^{-1}$  vibration bears a large intensity relative to the 2110  $\text{cm}^{-1}$  vibration.

## CRYSTAL STRUCTURE

The structure of  $[\text{Ni}(\text{pn})_2]_2[\text{Fe}(\text{CN})_6]\text{ClO}_4 \cdot 2\text{H}_2\text{O}$  (4) has been preliminarily reported.<sup>9</sup> The ORTEP drawing of the essential part of the crystal structure is given in Figure 1. The asymmetric unit comprises of each one-half of  $[\text{Fe}(\text{CN})_6]^{3-}$  and  $\text{ClO}_4^-$  ions, one  $[\text{Ni}(\text{pn})_2]^{2+}$ , and one water molecule. Hexacyanoferrate(III) ion makes bonds to four  $[\text{Ni}(\text{pn})_2]^{2+}$  ions through its four cyano nitrogens on a plane, providing a 2-D network extended through Fe(III)-CN-Ni(II) linkage (see Figure 2). The mean Fe-C, C-N, and N-Ni bond lengths are 1.949, 1.142, and 2.105 Å, respectively. The intrasheet Fe--Ni(1) and Fe--Ni(2) separations are 5.062(2) and 5.126(3) Å, respectively. The quadangle formed by four Fe ions shows a slight distortion to a rhombus with the interior angles of 85.0 and 95.0°. All the Ni ions assume a *trans* octahedral geometry

together with two cyanide nitrogens at the axial sites. Perchlorate ion resides within each quadrangle.

In the lattice the 2-D sheets align along *c* axis with the Fe--Fe(0, 1/2, 1) separation of 8.978(5) Å. The nearest intersheet Fe--Ni(1)(*x*, *y*, *z*+1), Fe--Ni(2)(*x*, *y*+1, *z*+1) and Ni(1)--Ni(2)(*x*+1, *y*+1, *z*+1) separations are 9.852, 8.613, and 9.754 Å, respectively.

Based on our preliminary studies, the 1,1-dmen compounds are shown to have a 2-D network essentially the same as that of **4**. In spite of many efforts we have not yet succeeded in growing good crystals suitable for X-ray crystallography for the en compounds.

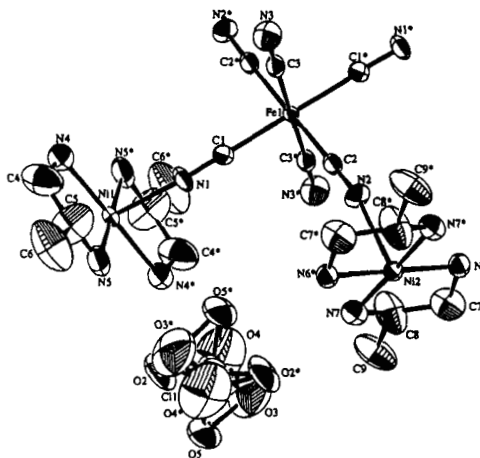


FIGURE 1 ORTEP drawing of the asymmetric unit of **4** with atom numbering scheme (water molecules are omitted for clarity).

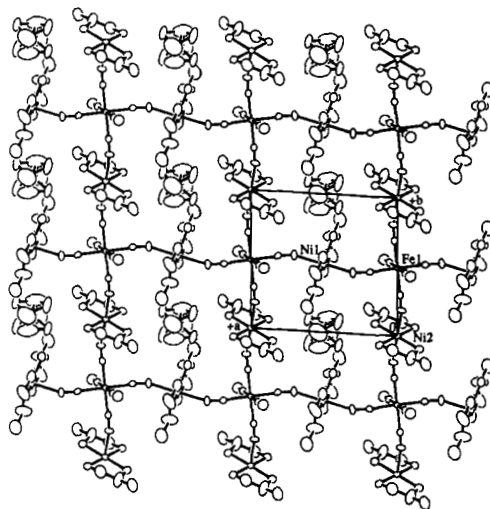


FIGURE 2 Projection of the polymeric structure of **4** onto the *ab* plane.

## MAGNETIC PROPERTY

### En Compounds (1 - 3)

Cryomagnetic properties of the en compounds have been studied in the temperature range of 4.2-90 K. The  $\chi_M T$  vs.  $T$  and  $\chi_M$  vs.  $T$  plots of 1 are exemplified in Figure 3.

The  $\chi_M T$  value at room temperature is  $2.90 \text{ cm}^3 \text{ K mol}^{-1}$  ( $4.81 \mu_B$ ) per  $\text{FeNi}_2$ , that is slightly larger than the value expected for one low-spin  $\text{Fe(III)}$  and two paramagnetic  $\text{Ni(II)}$  ions ( $2.38 \text{ cm}^3 \text{ K mol}^{-1}$ ;  $4.36 \mu_B$ ). The  $\chi_M T$  value increases with decreasing temperature, reaches the maximum value of  $6.09 \text{ cm}^3 \text{ K mol}^{-1}$  ( $6.98 \mu_B$ ) at 14 K, and then decreases below this temperature. The maximum  $\chi_M T$  value is significantly larger than the value expected for  $S_T = 5/2$  of ferromagnetically coupled  $\text{Ni(II)Fe(III)Ni(II)}$ . This fact suggests an intramolecular ferromagnetic interaction through the cyanide bridge. This is supported by a large Weiss constant ( $\theta = +9.5 \text{ K}$ ) determined by the Curie-Weiss plots in the range of 13 - 90 K. The drop in  $\chi_M T$  below 9 K suggests the operation of an intermolecular antiferromagnetic interaction.

The  $\chi_M$  vs.  $T$  plot shows a maximum at 14 K and a minimum at 6 K. Below this temperature a tendency of saturation is seen. The behavior in  $\chi_M$  at low temperature is typical of antiferromagnets and the non-zero  $\chi_M$  value can be attributed to a residual  $\chi_1$  component. No spontaneous magnetization occurs in 1 - 3.

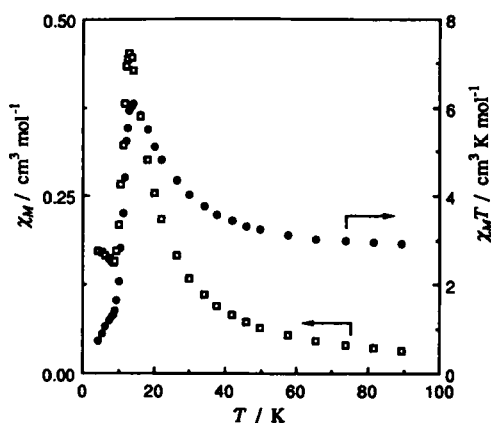


FIGURE 3 Temperature-dependences of  $\chi_M$  ( $\square$ ) and  $\chi_M T$  ( $\bullet$ ) per  $\text{FeNi}_2$  unit of 1.

### Pn Compounds (4 - 6)

All the pn compounds are similar to each other in magnetic behavior. The cryomagnetic property of 4 is shown in Figure 4 in the forms of  $\chi_M T$  vs.  $T$  and  $\chi_M$  vs.  $T$  plots. At room temperature the  $\chi_M T$  value is  $3.21 \text{ cm}^3 \text{ K mol}^{-1}$  per  $\text{FeNi}_2$  ( $5.07 \mu_B$ ) which is

larger than the expected value for one low-spin Fe(III) and two paramagnetic Ni(II) ( $2.38 \text{ cm}^3 \text{ K mol}^{-1}$ ;  $4.36 \mu_B$ ). With decreasing temperature the  $\chi_M T$  increases up to the maximum value of  $11.31 \text{ cm}^3 \text{ K mol}^{-1}$  ( $9.51 \mu_B$ ) near 10 K and then decreases below this temperature. The result clearly indicates a ferromagnetic ordering of spins within the 2-D sheet through the Fe-CN-Ni linkages and the drop in  $\chi_M T$  below 10 K can be ascribed to an antiferromagnetic intersheet interaction. The  $\chi_M$  vs.  $T$  curve shows a maximum at 10 K and a minimum at 6 K. The non-zero value of  $\chi_M$  below 6 K can be attributed to a residual  $\chi_1$  component of antiferromagnets.

The increasing tendency of  $\chi_M$  at low temperature suggests the magnetic ordering over the lattice. To confirm this, the magnetization  $M$  was studied in the temperature range of 4.2 - 20 K under a weak applied field 3 G (Figure 4). The field-cooled magnetization (FCM) curve showed a peak near 10 K and a break for saturation near 8 K. Essentially the same FCM curve was obtained under 100 G. When the applied field was switched off at 4.2 K, a remnant magnetization (RM) was observed which decreased with increasing temperature and vanished at 8 K. The zero-field-cooled magnetization (ZFCM) curve showed two maxima at 8.0 and 9.9 K which correspond to the breaking point and the maximum in the FCM curve, respectively. Such cryomagnetic behavior is typical of metamagnetic ordering and the peak at 8.0 K implies that the applied field is too weak to move the domain walls below this temperature.

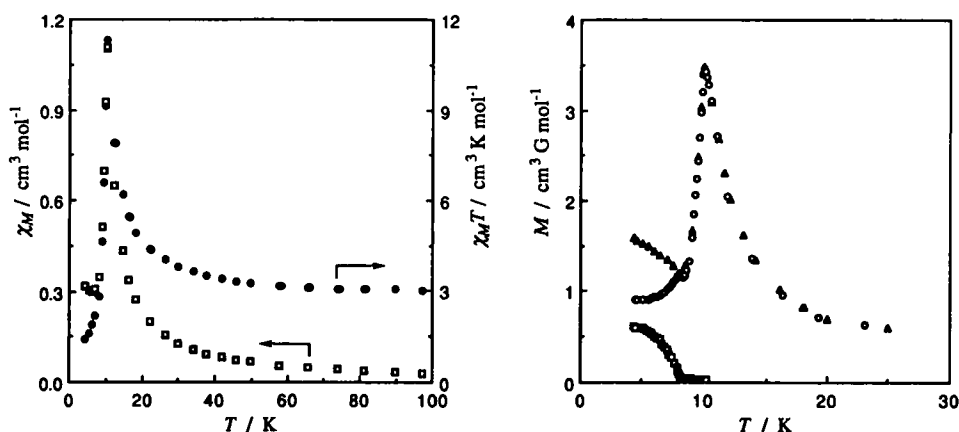


FIGURE 4 Temperature-dependences of  $\chi_M$  ( $\square$ ) and  $\chi_M T$  ( $\bullet$ ) per  $\text{FeNi}_2$  unit of **4** (left) and the magnetization  $M$  for **4** under an applied field of 3 G (right): FCM ( $\blacktriangle$ ), RM ( $\square$ ), ZFCM ( $\circ$ ).

The magnetic hysteresis loop determined at 4.2 K also adds a support to metamagnetic nature of this compound (Figure 5). The magnetization increased abruptly

around 3800 G, probably due to spin flipping based on the intersheet antiferromagnetic interaction. The magnetization was  $2.47 \mu_B$  under 20 kG and was not saturated under 80 kG ( $3.88 \mu_B$ ). It appears that each local spin is canted and a large magnetic field is necessary for overcoming the canting.

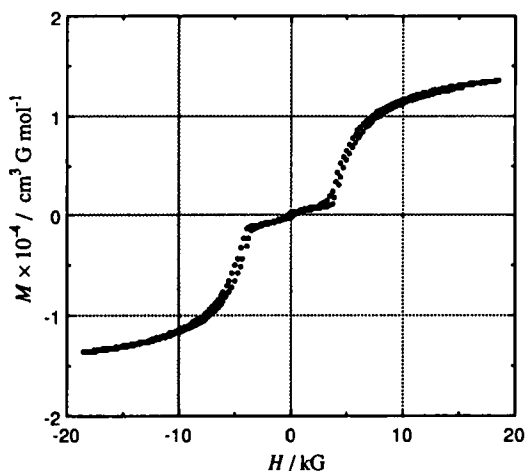


FIGURE 5 Magnetic hysteresis loop of **4** at 4.2 K.

### 1.1-Dmen Compounds (**7** - **9**)

The 1,1-dmen compounds are suggested to have a 2-D network structure like **4**, based on our preliminary X-ray crystallographic studies. In fact the 1,1-dmen compounds **7** - **9** resemble the pn complexes **4** - **6** in magnetic property. The  $\chi_M T$  vs.  $T$  and  $\chi_M$  vs.  $T$  curves of **7** are shown in Figure 6. The  $\chi_M T$  value at 90 K is  $3.38 \text{ cm}^3 \text{ K mol}^{-1}$  per  $\text{Ni}_2\text{Fe}$  ( $5.20 \mu_B$ ) that increases with decreasing temperature up to the maximum value of  $32.58 \text{ cm}^3 \text{ K mol}^{-1}$  ( $16.1 \mu_B$ ) at 11 K. The  $\chi_M$  vs.  $T$  curve shows a maximum at 11 K, a minimum at 7 K, and a trend to increase below 7 K. It appears that in **7** the intrasheet ferromagnetic interaction is strengthened whereas the intersheet antiferromagnetic interaction is weakened relative to the case of **4**. The reduced intersheet interaction in **7** may be due to the large intersheet separation relative to that of **4**, owing to the bulkyness of the diamine 1,1-dmen.

The temperature-dependence of the magnetization of **7** was studied under a weak applied field 3 G (see Figure 6). The FCM curve shows a peak at 11 K and a break for saturation near 10 K. A remnant magnetization is observed in the temperature range below 12 K. The ZFCM curve shows a maximum at 11.1 K. Thus, a long-range magnetic ordering occurs in this compound ( $T_C = 11.1 \text{ K}$ ). Similarly, **8** and **9** show a magnetic ordering at 14.6 and 11.0 K, respectively.

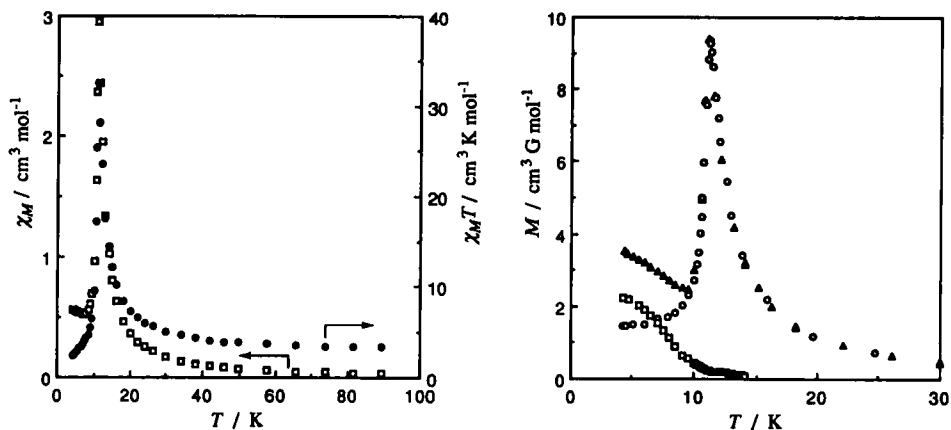


FIGURE 6 Temperature-dependences of  $\chi_M$  ( $\square$ ) and  $\chi_M T$  ( $\bullet$ ) per FeNi<sub>2</sub> unit of **7** (left) and the magnetization  $M$  for **7** under an applied field of 3 G (right): FCM ( $\blacktriangle$ ), RM ( $\square$ ), ZFCM ( $\circ$ ).

The magnetic hysteresis loop of **7** is given in Figure 7, which shows an abrupt increase of magnetization around *ca.* 1800 G. Thus, **7** is metamagnetic in nature. The spin flipping in **7** occurs under a lower magnetic field (*ca.* 1800 G) compared with **4** (*ca.* 3800 G). The magnetization of **7** was practically saturated under 80 kG (4.96  $\mu_B$  per FeNi<sub>2</sub>). It appears that the intersheet antiferromagnetic interaction is weak in **7**, relative to **4**, probably because of the large intersheet separation in the former. The remnant magnetization and the coercive field of **7** are 1140 cm<sup>3</sup> G mol<sup>-1</sup> and 600 G, respectively.

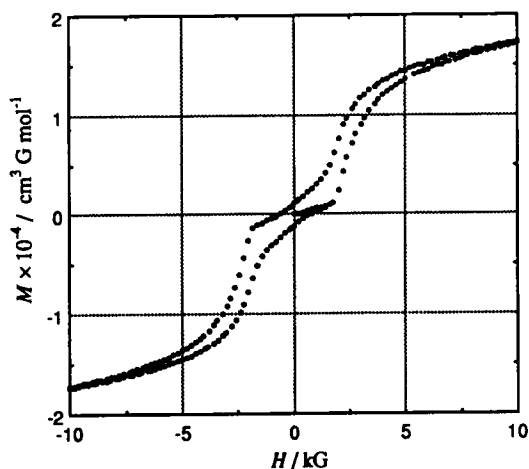


FIGURE 7 Magnetic hysteresis loop of **7** at 4.2 K.



## CONCLUSION

Bimetallic assemblies of the general formula  $[\text{Ni}(\text{diamine})_2]_2[\text{Fe}(\text{CN})_6]\text{X}$  (diamine = en, pn, 1,1-dmen);  $\text{X}^- = \text{ClO}_4^-$ ,  $\text{BF}_4^-$  or  $\text{PF}_6^-$ ) have been prepared and studied in their magnetic property. The en compounds of unknown network structure show a ferromagnetic spin-exchange between the adjacent Fe(III) and Ni(II) ions but no long-range magnetic ordering. The pn and 1,1-dmen compounds have a 2-D network of square structure and show a long-range magnetic ordering ( $T_C$ : 9.4 - 14.6 K).

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